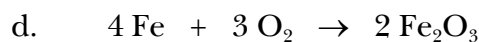
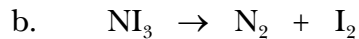
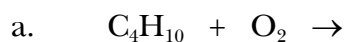


CHEMISTRY 3A --- WORKSHEET 9

CHAPTER 6

1. For each of the following reactions, show the products that would be expected to form and balance the equation. Be sure to label the state of each substance, e.g. (s), (aq), etc. If no reaction would occur, simply write N.R. Also label each reaction as to type, e.g. single displacement, combustion, decomposition, or combination.



2. Identify the oxidizing agent, reducing agent, substance oxidized, and substance reduced in each of the following reactions.

